

Chemistry

<u>Course Description</u>: Chemistry is the study of the composition of matter and the changes matter undergoes. Chemistry provides the groundwork for understanding in all other branches of science. Students will establish comprehensive knowledge through many in-class activities, guided practice and labs. The course will establish a strong foundation for further study in all sciences.

Attendance: Students are required to be in school every day. Students are responsible for communicating with their teachers to make up for missed learning.

<u>Course Expectations:</u> Although not everything is graded, everything is important. In order to demonstrate growth and learning, students will need to:

- 1. Participate in class activities (take notes, work in a group, complete in class tasks, ask questions) without distractions (cell phones, games, etc.)
- 2. Use morning time and the teacher to seek help outside of class when needed.
- 3. Complete all assessments within teacher timelines.

Grading

Skyward Assessment (Grades) may include quizzes, labs, learning checks, tests, speeches, performances, and projects.

Final Grade							
Α	100% - 90%						
В	80% - 89%						
С	70% - 79%						
D	60% - 69%						
F	0% - 59%						

Calculations:

Final Grade Calculation							
Cumulative Grade	95%						
Final Exam(s)	5%						

Instructional Resources:

- Canvas
- periodicvideos.com(daily element videos)
- youtube (scientific videos)
- ck12 (makeup labs)

Course Calendar/Pacing:

_August/September						October							November								
21 22 23 24 25 26						1	1 2 3 4 5 6 7							1 2 3 4							
27	28	29	30	31	1	2	8		10	11	12	13	14	5	6	7	8	9	10	11	
3	4	5	6	7	8	9	1!	16	17	18	19	20	21	12	13	14	15	16	17	18	
10	11	12	13	14	15	16	2:	2 23	24	25	26	27	28	19	20	21	22	23	24	25	
17	18	19	20	21	22	23	29							26	27	28	29	30			
24	25	26	27	28	29	30															
		De	ecemb	er				January							February						
					1	2		1	2	3	4	5	6					1	2	3	
3	4	5	6	7	8	9	7	8	9	10	11	12	13	4	5	6	7	8	9	10	
10	11	12	13	14	15	16	14	15	16	17	18	19	20	11	12	13	14	15	16	17	
17	18	19	20	21	22	23	2:	. 22	23	24	25	26	27	18	19	20	21	22	23	24	
24	25	26	27	28	29	30	28	29	30	31				25	26	27	28	29			
	March					April							May								
					1	2	33		2	3	4	5	6				1	2	3	4	
3	4	5	6	7	8	9	7		9	10	11	12	13	5	6	7	8	9	10	11	
10	11	12	13	14	15	16	14		16	17	18	19	20	12	13	14	15	16	17	18	
17	18	19	20	21	22	23	2:		23	24	25	26	27	19	20	21	22	23	24	25	
24	25	26	27	28	29	30	28	29	30					26	27	28	29	30	31		
	Unit 1					Unit 2							Unit 3								
	Unit 4				Unit 5							Unit 6									
	Unit 7						Unit 8							Unit 9							
	Unit 10					Semester Exams															
	No School				Pr	Professional Development / Teacher Workday							Early Release								

		Expected Pacing				
Unit 1 – Atomic Structure	5 weeks	Semester 1				
Unit 2 – Electron Configuration	2 weeks					
Unit 3 – Periodic Table	3.5 weeks					
Unit 4 - Chemical Bonding	4 weeks					
Unit 5 – Mole	3.5 week					
Unit 6 – Stoichiometry	4 weeks					
Unit 7 - Chemical Reactions	3 weeks					
Unit 8 - Acids & Bases	3 weeks	Semester 2				
Unit 9 – Solutions	3 weeks					
Unit 10 - Nuclear Chemistry	2.5 weeks					

^{*}Keep in mind that everything is tentative and subject to change